

Prophylaxis with α -lipoic acid against lipopolysaccharide-induced brain injury in rats

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Abstract

Introduction: Lipopolysaccharide (LPS) stimulates the synthesis and release of reactive oxygen species that play an important role in the pathogenesis of tissue injuries. In this study the effect of early administration of the antioxidant α -lipoic acid (α -LA) on brain lipid peroxidation, brain hydrogen peroxide (H_2O_2) concentration, and brain total sulfhydryl group (-SH group) content was evaluated in rats with endotoxic shock induced by administration of LPS (*Escherichia coli* 026:B6, 30 mg/kg i.v.)

Materials and Methods: Rats were treated intravenously with normal saline or α -LA (60 mg/kg) 30 min after LPS injection. After 5 h of observation, the animals were killed and their brains were isolated for the measurements.

Results: Injection of LPS alone resulted in the development of shock and oxidative stress, the latter indicated by a significant increase in brain concentrations of thiobarbituric acid-reacting substances (TBARS) and H_2O_2 and a decrease in total brain -SH group content. Administration of α -LA after the LPS challenge resulted in an increase in total -SH group content and a decrease in TBARS and H_2O_2 concentration in the brain tissue compared with the LPS group.

Conclusion: The results indicate that α -LA treatment effectively protected the brain tissue against endotoxin-induced oxidative stress. Administration of LA could be a useful adjunct to clinical application in the management of septic shock.

Key words: lipopolysaccharide, alpha-lipoic acid, oxidative stress.

Abbreviations: LPS – lipopolysaccharide, LA – lipoic acid, H_2O_2 – hydrogen peroxide, TBARS – thiobarbituric acid reactive substance, TNF- α – tumor necrosis factor α , IL – interleukin, ROS – reactive oxygen species, GSH – glutathione, GSSG – oxidized glutathione, -SH group – sulfhydryl group, TBA – thiobarbituric acid, TCA – trichloroacetic acid, BHT – butylated hydroxytoluene, HVA – homovanilic acid, PBS – phosphate-buffered saline, EDTA – ethylenediamine tetraacetic acid.

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INTRODUCTION

Lipopolysaccharide (endotoxin; LPS) is a major component of the outer membrane of Gram-negative bacteria and is responsible for the initiation of the host response to the microorganisms. The first phagocytic cells that come into contact with the LPS are macrophages and neutrophils. In response to LPS, macrophages secrete proinflammatory cytokines such as tumor necrosis factor (TNF)- α , the interleukins IL-6, IL-1 β , and IL-12 (Kagan et al. 1992; Panigrahi et al. 1996; Rankine et al. 2006), free radicals, and reactive oxygen species (ROS), such as superoxide anion (Minuz et al 2006; Victor et al 2003). An excess of ROS may cause cell death by oxidizing proteins, DNA damage, or lipid peroxidation of cellular

membranes (Floyd et al. 2001; Laskin and Pendino 1995). It is well documented that proinflammatory cytokines and ROS contribute to the development of septic shock, multiple organ failure, and death.

Brain tissue is highly susceptible to oxidative injury compared with other tissues (Halliwell 2006). The vulnerability of the brain to oxidative damage is probably due to membrane phospholipids rich in polyunsaturated fatty acids, which are a source of peroxidation, a high consumption of oxygen and a low quantity of enzymes mopping up ROS, and neurons' inability to multiply. As the capacity of the endogenous antioxidative system of brain tissue is limited, antioxidants such as LA could protect neural tissue from being damaged by increased ROS.

DL- α -lipoic acid (LA, DL-6,8-dithiooctanoic acid) is an endogenous thiol antioxidant that has the potential to quench ROS, regenerate glutathione (GSH), and chelate metals such as iron, copper, mercury, and cadmium, which are known to mediate free-radical damage in biological systems (Akpınar et al. 2007; Applegate et al. 2008; Bilska and Włodek 2005; Savitha et al. 2005; Skibska et al. 2006). As LA is well soluble not only in water, but in fats as well, and is a small molecule (molecular weight: 206), it may easily cross the blood-brain barrier and stabilize it (Cakatay 2006; Roy and Packer 1998; Roy et al. 1997; Schreibelt et al. 2006).

In most of the experiments performed by other authors, LA was administered before oxidative stress was observed. The present study was undertaken to investigate whether LA affects the level of lipid peroxidation, hydrogen peroxide (H_2O_2) concentration, and the content of sulfhydryl groups (-SH groups) in the brains of endotoxin-pretreated male rats.

MATERIALS AND METHODS

Chemicals

Lipopolysaccharide (*Escherichia coli* LPS 026:B6, lyophilized powder chromatographically purified by gel filtration, protein content <1%), α -LA, thiobarbituric acid (TBA), butylated hydroxytoluene (BHT), and sodium acetate trihydrate were purchased from Sigma Chemical Co. (St. Louis, MO, USA). All other reagents were obtained from POCh (Gliwice, Poland) and were of analytical grade. Shortly before use, the LPS was dissolved in sterile pyrogen-free normal saline. α -LA was mixed with sterile normal saline in a dark bottle and NaOH was added until the suspension was dissolved. The pH was then brought to 7.4 with HCl. TBA solution was prepared by dissolving 0.67 g of TBA in 100 ml of de-ionized water and then diluted 1:1 with glacial acetic acid. Sterile, de-ionized water (resistance >18 M Ω -cm, HPLC Water Purification System USF ELGA, England) was used throughout the study.

Animals

The experiments were performed on 52 adult male Wistar rats weighing 260–300 g. The animals were housed six rats per cage under standard laboratory conditions, with a 12/12 h light-dark cycle (light on at 7:00 a.m.), an ambient temperature of $20 \pm 2^\circ\text{C}$, and humidity of $55 \pm 5\%$. All the animals received a standard laboratory diet and water *ad libitum*. All the animals were given a one-week acclimation period before the start of the experiment. The experimental procedures followed the guidelines for the care and use of laboratory animals and were approved by the Medical University of Łódź Ethics Committee.

Experimental design

The animals were randomly divided into five groups. Group I (n=10, NaCl/NaCl control rats) received 0.2 ml saline twice, 0.5 h apart. Group II (n=10, NaCl/LA60 group) served as LA controls; the animals received 0.2 ml saline and were injected 0.5 h later with a single dose (60 mg/kg b.w.) of DL- α -LA. Group III (n=10, NaCl/LA100 group) received 0.2 ml saline and were injected with a single dose (100 mg/kg) of DL- α -LA 0.5 h later. Group IV (n=12, with 2 dead animals during the experiment, NaCl/LPS group) served as LPS controls; the rats were given 0.2 ml saline and endotoxic shock was induced 0.5 h later by an injection of *E. coli* LPS at a concentration of 30 mg/kg. Group V (n=10, LPS/LA60 group) received a single dose of LPS (0.2 ml, 30 mg/kg) and a single dose of LA (60 mg/kg) after 0.5 h. Group VI (n=10, LPS/LA100 group) received a single dose of LPS (0.2 ml, 30 mg/kg) and a single dose of LA (100 mg/kg) after 0.5 h.

All agents were injected intravenously into the tail vein between 8:00 and 9:00 a.m.. The rat's body temperature was taken in the rectum with a thermistor thermometer (type PU 319/1, Czechoslovakia) before the first injection and then 1, 2, 3, and 5 h later. After 5 h of observation, the animals were anesthetized with pentobarbital (50 mg/kg i.p.).

Brain homogenates

The skulls were split on an ice and salt mixture and the whole brain of each rat was extracted, weighed, and homogenized in either 1.15% KCl for the assay of TBA reactive substance (TBARS) and H_2O_2 concentration or in 6% trichloroacetic acid (TCA) for determination of the content of free -SH groups.

Determination of lipid peroxidation

The content of lipid peroxidation products in the brain homogenates was assayed as TBARS. Briefly, 4 ml of 0.25% hydrochloric acid containing 0.375% TBA, 15% TCA, and 0.015% BHT was added to 2 ml of the brain homogenate. The samples were boiled for 30 min at 100°C in tightly closed tubes. After cooling to 10°C , 2.5 ml of butanol was added to each tube and, after intensive shaking, the tubes were centrifuged for 10 min (3800 rpm, 20°C). TBARS in the butanol layer were measured spectrofluorometrically using a Perkin Elmer LS-50 Luminescence Spectrometer (Norwalk, CT, USA). Excitation was set at 515 nm and emission was measured at 546 nm. The readings were converted into μM by the regression equation

$$Y = 0.39 \times (X - X_0) - 1.32,$$

where Y – $\mu\text{mol/l}$ TBARS and X and X_0 the fluorescence intensities of the sample and control, respectively, in arbitrary units (arbitrary units, AU). The regression equation was prepared from three series of calibration experiments

with six increasing concentrations of tetramethoxypropane as a standard of TBARS (0.01–50 μ M). A mixture of 2 ml of 1.15% potassium chloride and 4 ml of 0.25 N hydrochloric acid was used as a control. Finally, the results were calculated for 50 mg of brain tissue.

Determination of H_2O_2

H_2O_2 generation in the brain homogenates was determined according to the Ruch et al. method (Ruch et al. 1983) with some modifications. Briefly, a 10- μ l aliquot of tissue homogenate was mixed with (i) 90 μ l of phosphate-buffered saline (PBS, pH 7) and 100 μ l of horseradish peroxidase (1 U/ml) containing 400 μ mol homovanilic acid (HRP+HVA assay) or (ii) 90 μ l of PBS and 100 μ l of 1 U/ml horseradish peroxidase only (HRP assay). Both homogenates were incubated for 60 min at 37°C. Then 300 μ l of PBS and 125 μ l of 0.1 M glycine-NaOH buffer (pH 12.0) with 25 mM EDTA were added to each homogenate sample. Excitation was set at 312 nm and emission was measured at 420 nm (Perkin Elmer Luminescence Spectrometer, Beaconsfield, UK). The readings were converted into H_2O_2 concentration using the regression equation

$$Y = 0.012 \times X - 0.0007,$$

where Y is the concentration of H_2O_2 in the homogenate (μ M) and X the intensity of light emission at 420 nm for the HRP+HVA assay reduced by the HRP assay emission (AU). The regression equation was prepared from three series of calibration experiments with 10 increasing H_2O_2 concentrations (range: 10–1000 μ M). The lower limit of H_2O_2 detection was set 0.1 nM, with intra-assay variability not exceeding 2%.

Measurement of the total -SH group content in brain homogenates

The -SH group content in the tissue homogenates was determined by the 2,2-dithiobisnitrobenzoic acid (DTNB) assay described by Ellman (Ellman 1970).

Briefly, 50 mg of the organ sample was thawed and homogenized with 6% TCA in a 1:20 volume proportion at 0°C. Then the homogenate was centrifuged for 20 min (10,000 \times g, 4°C). The volume of 0.5 ml of the supernatant was mixed with 0.5 ml of 0.3 M Na_2HPO_4 and 0.5 ml of 0.04% Ellman reagent (DTNB) freshly dissolved in 10% sodium citrate water solution. All the reagents added were previously cooled to 0°C. The absorbance of the obtained solution was measured at 412 nm using a Pharmacia LKB-Ultrospect III spectrophotometer. The readings were converted into -SH group content (μ M) by the regression equation

$$Y = -0.2998 + 241.954 \times X,$$

where Y is the free -SH group content (μ M) and X the intensity of light emission at 420 nm (AU). The regression equation was prepared on the basis of three series of calibration experiments in which increasing concentrations of GSH (2–200 μ mol/l) were used. The control assay was carried out with a solution of 0.5 ml of 6% TCA, 0.5 ml of 0.3 M Na_2HPO_4 – sodium phosphate, and 0.5 ml of 0.04% DTNB.

Statistical analysis

Data are presented as the mean \pm SE from 10 animals in each group. Statistical significance was evaluated using the ANOVA test followed by Duncan's multiple range test as a post hoc test. Differences were considered significant if $p \leq 0.05$.

RESULTS

Body temperature and general behavior

In the control group there was no change in body temperature throughout the experiment ($33.2 \pm 0.1^\circ\text{C}$). Similarly, infusion of α -LA had no effect on body temperature. Injection of LPS resulted in a body temperature decline of about 2°C compared with the saline and Na/LA groups ($p < 0.001$). Some decrease in body temperature was observed in the LPS/LA60 group, but after 5 h of

Table 1. The effect of lipoic acid on LPS-induced changes in the body temperature in rats

Temperature/ Group of animals	Before	1 h	2 h	3 h	5 h
NaCl/NaCl	33.2 \pm 0.20	33.2 \pm 0.20	33.2 \pm 0.20	33.2 \pm 0.20	33.2 \pm 0.20
NaCl/LA60	33.7 \pm 0.20	33.8 \pm 0.25	33.7 \pm 0.30	33.8 \pm 0.20	33.5 \pm 0.20
NaCl/LA100	33.5 \pm 0.20	33.6 \pm 0.20	33.4 \pm 0.23	33.3 \pm 0.20	33.3 \pm 0.15
NaCl/LPS	33.3 \pm 0.13	32.0 \pm 0.19 ^b	31.6 \pm 0.20 ^b	31.5 \pm 0.20 ^b	31.5 \pm 0.20 ^b
LPS/LA60	33.3 \pm 0.15	33.0 \pm 0.29	32.2 \pm 0.30 ^a	32.2 \pm 0.40 ^a	32.6 \pm 0.30
LPS/LA100	33.3 \pm 0.20	33.2 \pm 0.20	33.2 \pm 0.20	33.0 \pm 0.29	33.1 \pm 0.22

LPS (30 mg/kg, i.v.) or 0.2 ml saline were administered into the tail vein and the rats were injected 0.5 h later with LA (60 or 100 mg/kg, i.v.). Body temperature was measured in the rectum before and 1, 2, 3, and 5 h after injections. Mean \pm SE, ^a $p < 0.001$, ^b $p < 0.05$ vs. baseline temperature.

observation the body temperature rose to $32.6 \pm 0.3^\circ\text{C}$ (Table 1). In the NaCl/LA100 group, no decrease in body temperature was observed. Animals challenged with LPS alone displayed reduced motor activity and adopted a hunched posture. The changes in animal behavior were visible within 1 h after LPS injection. In contrast, the behavior of the animals of the LPS/LA60 and LPS/LA100 groups was similar to that of the control group. Mortality in the NaCl/LPS group was about 17%. No animals died in the control, NaCl/LA60, NaCl/LA100, and LPS/LA groups.

Changes in brain lipid peroxidation levels

The TBARS level in the brain homogenates after LPS administration (group IV) was 3.3-fold higher than in the control group ($p < 0.001$, $n = 10$). The level of TBARS in the NaCl/LA60 group slightly decreased ($p < 0.05$) compared with the control group. LA 60 mg/kg administered after LPS challenge (group V) significantly suppressed the extent of lipid peroxidation in the brain homogenates compared with the NaCl/LPS group (1.49 ± 0.2 vs. $4.37 \pm 0.37 \mu\text{M}$, $p < 0.001$, $n = 10$). Similar results were obtained with a 1.5-fold higher dose of LA (Table 2).

Changes in H_2O_2 concentration

The concentration of H_2O_2 in brain homogenate was significantly higher in the animals treatment with LPS than in the controls (0.173 ± 0.013 vs. $0.037 \pm 0.006 \mu\text{M}$, $p < 0.001$, $n = 10$). The administration of LA (60 mg/kg b.w.) after LPS challenge significantly reduced H_2O_2 concentration in the brain homogenates compared with the LPS group (0.113 ± 0.014 vs. $0.173 \pm 0.013 \mu\text{M}$, $p < 0.01$, $n = 10$). LA at a dose of 100 mg/kg b.w. reduced H_2O_2 concentration in the brain homogenates in the same range of values as LA at a dose of 60 mg/kg (Table 2).

Content of total -SH groups in the brain

Rats treated with LPS (group IV) had significantly ($p < 0.01$) lower -SH groups levels than rats treated with LA (groups II and III). Administration of LA (60 mg/kg b.w.) after LPS challenge significantly increased the

content of -SH groups in the brain homogenates compared with the NaCl/LPS group (5.53 ± 0.33 vs. $4.06 \pm 0.28 \mu\text{M}$, $p < 0.01$, $n = 10$). Similar results were obtained with the higher dose of LA (100 mg/kg b.w.; Table 2).

DISCUSSION

These data show that intravenous administration of LPS causes oxidative brain injury which is manifested by a significant increase in TBARS and H_2O_2 concentration and a decrease in the concentration of -SH groups in brain homogenates. The increase in lipid peroxidation in this study is a result of increased ROS production, which in turn leads to excessive peroxidation of polyunsaturated fatty acids, causing neuron damage (Halliwell 2006; Halliwell and Gutteridge 1997; Muthuswamy et al. 2006). It was previously reported that LPS, a potent inducer of inflammation and activator of macrophages, produces an abundance of ROS in microglia cells present in the brain (Halliwell 2006). The neurotoxic effects of LPS on brain neurons are mediated by the activation of microglial cells, IL-1 β , and caspase-11 (Arai et al. 2004). Moreover, the systemic administration of LPS (experimental model of sepsis) causes a release of glutamate, aspartate, and taurine in the central nervous system which in excess cause disturbances in calcium homeostasis, leading to the production of free radicals (Lin et al. 1999; Packer et al. 1997).

The increased formation of ROS in our study was indicated by the increased level of H_2O_2 in the brains of the LPS-challenged rats. H_2O_2 is an important component of the cascade of events during which ROS are produced (Halliwell 2006; Panigrahi et al. 1996) and is one of the most stable toxic oxygen metabolites. It is also volatile and, due to its lack of charge, easily crosses cell membranes by simple diffusion. This allows the promotion of radical reactions at a great distance from its origin. Catalase and GSH peroxidase are basic enzymes regulating intracellular H_2O_2 concentration. In the brain, catalase activity is low and the antioxidant enzyme superoxide dismutase is localized mainly in neurons (Delacourte et al. 1988), while GSH peroxidase and

Table 2. The influence of administration of LA on oxidative stress parameters in brain homogenates of rats subjected to LPS-induced shock

Parameter	Saline	LA60	LA100	LPS	LPS+LA60	LPS+LA100
TBARS (μM)	1.320 ± 0.022	0.870 ± 0.010^b	0.860 ± 0.040^b	$4.370 \pm 0.370^{a,b}$	$1.490 \pm 0.200^{a,d}$	$1.470 \pm 0.028^{a,c,d}$
H_2O_2 (μM)	0.037 ± 0.005	0.039 ± 0.006	0.036 ± 0.004	$0.173 \pm 0.013^{a,b}$	$0.113 \pm 0.014^{b,c}$	$0.111 \pm 0.013^{b,c,e}$
Free-SH groups (μM)	5.310 ± 0.270	5.510 ± 0.300	5.610 ± 0.270	$4.060 \pm 0.280^{a,c}$	5.530 ± 0.330^e	5.550 ± 0.290^e

Two doses of LA were used: 60 and 100 mg/kg b.w. Mean \pm SE.

All administrations were performed intravenously between 8:00 and 9:00 a.m. ^a $p < 0.001$ vs. LA60 or LA100, ^b $p < 0.001$ vs. saline, ^c $p < 0.01$ vs. saline, ^d $p < 0.001$ vs. LPS, ^e $p < 0.01$ vs. LPS.

Abbreviations: LPS – lipopolysaccharide, LA60 – lipoic acid 60 mg/kg body weight, LA100 – lipoic acid 100 mg/kg; TBARS – thiobarbituric acid-reactive substances.

GSH are mainly in astrocytes (Benzi and Moretti 1995). A deficit of these enzymes and their localization in the brain suggest that neural tissue may be especially susceptible to hydroxyl radicals (Packer et al. 1997). In case of an inadequate level of enzymes able to degrade H_2O_2 , more H_2O_2 can be converted to toxic hydroxyl radical that could then contribute to oxidative stress.

In this study, the decrease in the content of -SH groups in the LPS-treated rats may be a result of decreased activity of GSH reductase (Rankine et al. 2006) and transferase or increased activity of GSH peroxidase (Malmezat et al. 2000). A decreased concentration of tissue GSH and an increase in GSSG concentration during the initial phase of septic shock was observed by Tsiotou et al. (Tsiotou et al. 2005).

The present study also showed that LA at a dose of 60 mg/kg is highly effective in reducing free radicals and lipid peroxidation in the brains of septic rats. Administration of LA 0.5 h after i.v. LPS injection caused significant decreases in both TBARS and H_2O_2 concentrations in brain homogenates. However, more than a 1.5-fold increase in the dose of LA did not cause a further reduction in the scale of oxidative brain damage. Moreover, LA administration also resulted in improved survival rate and less advanced hypothermic reaction of septic rats compared with untreated animals, which is in agreement with other reports (Riedel et al. 2003). Panigrahi et al. (Panigrahi et al. 1996) indicated that the administration of LA to animals subjected to ischemia-reperfusion (e.g. by occlusion of a cerebral artery) alleviated the effects of reperfusion, lowered the level of ROS in the brain cells, reduced an extent of damage, and prolonged the survival time of the animals compared with a control group.

It was also demonstrated in the present study that early administration of LA after LPS challenge significantly increased the brain SH group content. An increase in -SH group concentration after administration of both LA doses (60 and 100 mg/kg b.w.) may indicate an enhanced synthesis of proteins containing SH groups or the synthesis of GSH. It has been demonstrated that LA increased the *de novo* synthesis of cellular GSH by improving cysteine utilization (Han et al. 1997). An increase in GSH concentration may lead to a higher activity of enzymes involved in GSH synthesis and a reduction in GSH as well (Malmezat et al. 2000). GSH maintains the reduced state of -SH groups required for the proper function of different types of proteins. For this reason, maintaining intracellular GSH at an appropriate level is essential for protecting neurons against oxidative stress. In *in vivo* and *in vitro* studies it has been shown that the administration of LA increased the intracellular GSH level by 30–70% (Han et al. 1995). Another action of α -LA in brain tissue and also in the other tissues is connected with its inhibition of NF- κ B and AP-1 (a dimer of c-fos and c-jun proteins) activity (Minuz et al 2006). Furthermore, LA contributes to other antioxidant systems by enhancing the effects of coenzyme Q10 and GSH (Savitha et al. 2005)

and also by regenerating other antioxidants, such as vitamins C and E (Kagan et al. 1992). All these effects of LA in LPS-induced sepsis are in accordance with the findings of LA's antioxidant properties. Moreover, this antioxidant can also inhibit acute inflammatory response *in vitro* and *in vivo* by activating the PI3/Akt pathway (Zhang et al. 2007).

In conclusion, this study suggests that LA limits damage to nerve tissue and protects it from being exposed to oxidizing stress even if this antioxidant is administered 0.5 h after LPS challenge. The effectiveness of LA's protective action against endotoxic brain damage indicated by our experiments could encourage the clinical use of this compound as a treatment for endotoxemia.

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